

Instrumental Analytical Chemistry INSTANCHEM

Part I

Self-assessment



Chemistry Degree

3rd Year

Dr. J.M. Fernández Álvarez



Sample INSTANCHEM test questions:

1. In UV-VIS spectroscopy, interaction of radiant energy with matter involves

- A. Nuclear configuration change
- B. Binding electron excitation
- C. Either molecular rotation or vibration

2. Isosbestic point of a system HA/A⁻ is one in which:

- A. pH = 2 pKa
- B. $\varepsilon_{HA} = \varepsilon_{A}$ -
- C. pH = pKa
- D. $[HA] = [A^-]$

3. If the molar absorptivity of HA is far superior to that of A⁻

- A. The calibration will be linear if done at pH = pKa
- B. The calibration will be linear if it is done in a strong acid medium
- C. The buffering of solutions will ensure the linearity of the calibration
- D. It is not possible to obtain a linear calibration

4. Beer's law is complied with:

- A. When the monochromator's bandpass is exactly 3 cm wide
- B. When the wavelengths used match the maximum of the absorption peak
- C. When the monochromator's bandpass is at least 10 times wider than the half-height absorption width of the analyte

5. In UV-VIS spectrometry, the cuvette must always be placed between the monochromator and the detector

- A. True
- B. False

6. Shot noise is due to

- A. Termal movement of electrons in the resistors
- B. Changes in temperatura in the conducting circuitry
- C. Movement of the electrons through a junction

7. The refractive index of the material used to manufacture a prism is to remain constant with wavelength

- A. True
- B. False

8. Indicate which of the following properties does not correspond to an ideal detector

- A. High sensitivity
- B. High signal / noise (S / N) ratio
- C. Long response time
- D. Absence of dark current

9. The simultaneous recording of a spectrum at all the wavelengths is possible with ...

- A. Time-resolved double beam spectrophotometer
- B. Space-resolved double beam spectrophotometer
- C. Photo Diode Array, PDA, detector

10. The S/N ratio

- A. Decreases when averaging spectra
- B. Increases with the square root of the the number of averaged spectra
- C. It is independent of the number of measuments done

11. An interference filter

- A. Provides a broader bandpass width than that provided by a colored filter
- B. Presents a lower Transmittance than a colored filter
- C. Is based on the phenomenon of scattering of radiation
- D. Shows bandwidths that are typically thinner than those of colored filters



12. The dispersion caused by a quartz prism

- A. It is linear along the focal plane of the exit slit
- B. It is larger in the red zone of the spectrum tan in the blue zone
- C. It is especially high in the UV

13. An echelette grating provides a uniform linear dispersion across the UV-VIS

- A. True
- B. False

14. Just considering the uncertainty in the measurement of the Transmittance

- A. To report the concentration, C, with a minimun error requires the measurements to be made in a range between 0.2 and 0.8 of absorbance units.
- B. The absolute value of T is irrelevant to the precise quantification of the analyte
- C. It is preferred to perform a calibration at such an interval of C that guarantees the minimum T
- D. It is demanded that absorbance must acquire a positive integer value > 1

15. The multiplicity of non-conjugated chromophores

- A. Exerts a marked influence on the wavelength
- B. Produces a hypsochromic effect
- C. Significantly increases the value of the molar absorptivity

16. Absorption by transition metal solutions

- A. Depends on the nature of the ligands present
- B. Gives sharp and well-defined signals
- C. Is independent of the nature of the solvent
- D. It always happen in the VIS regardless of the electronic structure in the layer d

17. In fluorescence and phosphorescence measurements, the detector is normally placed

- A. In line with the source
- B. Perpendicularly to the source
- C. Forming an angle of 135 degrees with the source

18. Fluorescence emissions

- A. Are obtained at wavelengths shorter than those corresponding to the respective absorption process
- B. They may involve excited states with different multiplicity to that of the ground state
- C. Occur between high vibrational levels of the first excited electronic state and the zero vibrational level of the electronic ground state
- D. Occur between the 0 vibrational level of the first excited electronic state and one of the vibrational levels associated with the ground electronic state

19. The lifetime of a triplet excited state is much lower than that of a singlet excited state

- A. True
- B. False

20. Stokes' shift

- A. Indicates a shift to higher energies of the fluorescent process with respect to the absorbing process
- B. Refers to a shift in the wavelength of the fluorescence to higher values with respect to the absorption wavelength
- C. It is not related to a radiational pathway energy dissipation process

21. The predissociation and dissociation phenomena are more favorable when energies absorbed from the incident beam give rise to transitions of the type:

- A. n-π*
- **Β**. σ-σ*
- **C**. π-π*

22. Quinine's fluorescence will be enhanced

- A. At high temperatura
- B. In the presence of 1-bromopropane
- C. In a micellar medium