

ICCC 2015 BEIJING

The 14th International Congress on the Chemistry of Cement

ABSTRACT BOOK Volume ||

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PREFACE

This book collects abstracts of the papers submitted to the 14th International Congress on the Chemistry of Cement (ICCC 2015), which was held in Beijing, China between October 13~16, 2015, and organized by China Building Materials Academy and the Chinese Ceramic Society. Since the First International Congress on the Chemistry of Cement started in London in 1918, it has become the most influential event for information exchange and promoting global cooperation on cement and concrete science and technology.

This congress attracted papers and attendee from 43 countries. It covers: (1) clinker chemistry, (2) hydration of portland cement, (3) supplementary cementing materials, (4) chemical admixtures and rheology, (5) durability and service life prediction, (6) alternative binders and (7) standard and codes. A large number papers discuss the considerable progress towards understanding chemistry, microstructure, and durability of cement and concrete. The challenge still remains of finding new ways to reduce CO₂ emission during cement and concrete production, to extend the service life of concrete. Considering cement and concrete as viable construction materials must be retained and sustained, considerable efforts still should focus on the environmental challenges that the industry and society now face. The main objective of this congress was to provide a platform for interactions between researchers, producers and engineers all over the world to produce cross-fertilization ideas aimed at developing rational miltiple-disciplinary ideas.

Caijun Shi and Yan Yao Editors

Overview of the Potential of Urban Waste Glass as A Cementitious	
Material in Alternative Chemically Activated Binders [100] and [100] and [100] and [100]	586
J Iván Escalante-Garcia	
Effect of Alkali Activator and Admixtures on The Properties of Plosphorous	
Slag-Based Cementitious Material	587
ZHANG Jun, PENG Yanzhou, XU Gang, LIU Jiuyan, KE Jin	
Preparation and Characteristics of Autoclaved Aerated Concrete Using	
Nickel Slag	588
Wu Qisheng Guang Jianmiao Zhu Huajun Zhang ChangsenZou Xiaotong	
Thermal and Mechanical Properties of Polyacrylonitrile Fiber	
Reinforced Fly Ash Based Foam Geopolymer	589
Ning-ning Shao, Ze Liu, Jian-jun Fan, Yuan Zhang, Dong-min Wang	W u m
BioCement - The Development of a CO ₂ -Neutral Cement	590
Natalie Carr, Lorenzo Tosti, Henk Jonkers	muga sel
Utilization of Thermal-treated Metallurgical Sludge as Cement Admixture	591
Xiaolu Guo, Liyan Zhang, Huisheng Shi	
Cementation of Simulated Liquid Nitrate Wastes by Alkali-activated	
Slag-metakaolin Cement and a statistical and a s	592
Nailia R. Rakhimova, Ravil Z. Rakhimov, Yury N. Osin, Natalia I. Naumkina, Alfiya M. Gubaidullina, Grigory I. Yakovlev, Arina V. Shaybadullina	gele l
Utilization of Fly Ash and Residual Rice Husk Ash for Producing	
Unfired Building Bricks with Cement Addition	593
Chao-Lung Hwang, Trong-Phuoc Huynh, Yogie Risdianto	
Effects of Foreign Elements on Anhydrous and Hydrous C2S Phases,	
A Possible Feedstock for Hydraulic Calcium Hydrosilicates	594
Pöhler, K., Pöllmann, H., Stemmermann, P., Garbev, K.	
Effect of Alkali-activated Solution Concentration on the	
Performance of Alkali-activated Slag Tunnel Fireproof Coatings	595
Deng-deng Zheng, Tao Ji, Wen-cheng Wang, Hwai-chung Wu	
Can We Make Sodium Carbonate-activated Slags React Faster?	596
Xinyuan Ke, Susan A. Bernal, John L. Provis	
Activation of Slags using Near-Neutral Salts: The Importance of Slag Chemistry	
Susan A. Bernal, Xinyuan Ke, John L. Provis	
Hydration Products And Microstructure Of Calcium Sulfoaluminate Cement	
Pastes In Different Curing Regimes	
F Liang Zhang, Fredrik P. Glasser	
Cements with A Low Clinker Content: Versatile Use of Raw Materials	
I. Garcia-Lodeiro, A. Fernández-Jimenez, A. Palomo	
The Role of Mgo Content in the Phase Assemblage of Alkali-Activated	
Synthetic Precursors within the System CaO-MgO-Al ₂ O ₃ -SiO ₂	
Brant Walkley, Rackel San Nicolas, Susan A. Bernal, John L. Provis, Jannie S. J. van Deventer	
Calcium Aluminate Cements as An Effective Matrix for Encapsulation of	
Hazardous Materials	601
I Navarro-Blasco R Sirera M Pérez-Nicolás A Duran IM Fernández II Alvarez	

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Calcium Aluminate Cements as An Effective Matrix for Encapsulation of Hazardous Materials

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Abstract

The use of calcium aluminate cements to efficiently encapsulate hazardous wastes was reported. Specifically, the quick formation of a matrix of low porosity and high stability was intended. The reaction between phosphate groups and calcium aluminate cement was extremely important in this case. This work presents an acid-base reaction between phosphates present in the sludges from the automotive industry and calcium aluminate cement that yielded very inert and stable monolithic blocks of amorphous calcium phosphate (ACP). Two industrial sludges of different compositions were characterized and loaded in different ratios (from 10 to 50 wt.%). Setting times and compressive strengths were recorded to establish the feasibility of this method to achieve a good handling and a safe landfilling of these samples. Short solidification periods were found and leaching tests showed an excellent retention for toxic metals such as Zn, Ni, Cu, Cr and Mn and also for organic matter. Retentions over 99.9% for Zn and Mn were observed even for loadings as high as 50 wt.% of the wastes. The formation of ACP phase of low porosity and high stability accounted for the effective immobilization of the hazardous components of the wastes, as confirmed by X-ray diffraction and SEM-EDAX studies. Comparison with the performance of pure sodium hexametaphosphate-CAC mortars bearing toxic metals solutions was sometimes reported.

Originality

The rationale is that polyphosphate-CAC matrices have shown interesting potential to solidify/stabilize heavy metals, owing to the aforementioned acid-base reaction that yields a compact and low porous matrix mainly composed of ACP - amorphous calcium phosphate -, which can be able to retain hazardous compounds. We aimed to take advantage of the reactivity of one of the sludge components: the sludges with relatively large concentrations of phosphate are expected to act themselves as reactants that, interacting with the CAC, could result in a very effective retaining system of the sludge constituents. Sludge samples from two locations have been incorporated in high proportion within the mix. We discussed the effects of the sludge on the CAC mortar and a possible interaction mechanism is provided.

Keywords: sludge; calcium aluminate cement; leaching; solidification/stabilization; landfilling



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prior to landfill disposal and cement is the most adaptable binder currently available for this immobilization. Polyphosphate-CAC matrices can turn out to be optimal candidates to solidify/stabilize heavy metals, owing to the acid-base reaction between acidic phosphate and calcium aluminate cement that yields a low porous matrix mainly composed of amorphous calcium phosphate (ACP), which can retain hazardous compounds. Thus, we expect the sludges with relatively large concentrations of phosphate act themselves as reactants, resulting – through their interaction with the CAC – in a very effective retaining system of the sludge constituents. This work is dealt with the encapsulation of two different automotive phosphate coating sludges in CAC-based mortars.

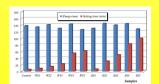
Material and methods

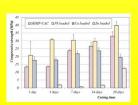
coming from the automotive industry. A control group with sodium hexametaphosphate (SHMP, (NaPO₃)_e), as a pure acidic phosphate reactant, was also prepared for comparison purposes. Table 1 collects the sludge waste amounts used and the required mixing water to achieve the consistency of the control sample.

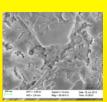
Different usual cement characterization techniques were applied. The leachability of the raw sludge wastes by water: EN 12457-4 extraction test method. The leaching performance of the monolithic specimens: semi-dynamic Tank Test (Concentration levels of the elements and the organic matter were measured by AAS and TOC, respectively).

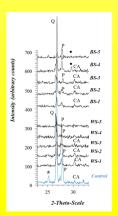
Characterization of the automotive sludges

Table 1 shows the larger the percentage of sludge, the higher the required mixing water. The presence of the automotive sludges under study here induced a delay in the setting time (Fig. 2), associated to the toxic metals presence in the sludges (Cu and Zn, specially), which inhibit the setting process. A reduction of the compressive strength takes place when either WS or BS was added to the mortar as compared with the control sample (Table 3). The large mechanical strengths of SHMP-CAC mortars (Fig. 3, higher than 30 MPa after 28 curing days) point to the stability and low porosity of the binding matrix, confirmed by SEM analysis (Fig. 4). The incorporation of the WS also promoted the formation of ACP, the presence of phosphates inhibited the hydration of CAC and no reflection.









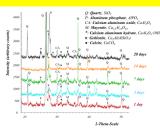
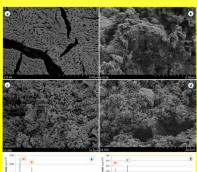
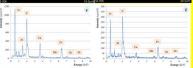
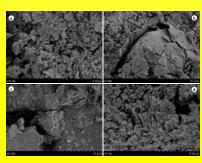


Figure 6. SHMP-CAC XRD patterns vs. curing times

Tal	ble 3 Comp	ressive	strengths	(CS) in	MPa for	PS-CAG	morta	s at 7 an	d 28 curi	ng days	
Raw material	Control	WS1	WS2	WS3	WS4	WS5	BS1	BS2	BS3	BS4	BS5
CS 7 days	25	18	10	7	7	6	12	6	4	6	5
CC 20 4	42	22	1.2	10			1.1	7	2	-	-







sample; b) BS5 sample; c) BS1 sample with small hexagonal platelets of CAH₁₀; d) BS5 with fibrous ACP

The appearance of this matrix can be reasonably associated to the ACP phase, in accordance with the XRD results.

The incorporation of BS generated a matrix with larger pores (Fig. 8a and 8b). At higher magnification micrographs showed CAH₁₀ hexagonal plate-like crystals into

matrix (Fig. 8c) as well as the fibrous network (Fig. 8d) in a lesser extent than in WS samples

Element	LLV	WS	WS	BS	BS	
Element	LLV	H ₂ O	HCl 0.1M	H ₂ O	HCl 0.1M	
TOC (mg L ⁻¹)	320	47.4±0.1	-	427.7±0.1	-	
Cr (mg L ⁻¹)	15	0.03±0.01	0.32±0.03	0.004±0.001	0.03±0.01	
Cu (mg L-1)	60	0.49 ± 0.02	0.61±0.07	0.29±0.01	0.29±0.01	
Mn (mg L-1)		79±6	333±10	0.33±0.03	13.9±0.7	
Ni (mg L ⁻¹)	12	82±4	217±9	2.14±0.01	18±1	
Zn (mg L ⁻¹)	60	243.7±0.7	1341±7	0.40±0.06	0.78±0.02	

Element	WS1	WS5	BS1	BS5
Cr	100.0	97.68	96.19	99.86
Cu	99.59	99.92	98.13	97.03
Mn	100.0	100.0	100.0	100.0
Ni	99.42	99.86	99.85	99.97
Zn	99.97	99.99	99.97	99.99
PO ₄ 3-	99.82	99.97	99.95	99.94

test in monorium specimens					
	WS1	WS5	BS1	BS5	
TOC	4.8 ± 0.1	2.0 ± 0.1	3.3 ±0.1	5.7 ± 0.1	



Conclusions

Leaching tests indicated that for the two sludge wastes, toxic metals such as Zn. Ni. Cr. Cu and Mn were nearly fully retained within the hinding matrix together with the largest fraction of the organic matter. The matrix has been proved to be efficient to successfully solidify/stabilize these has

Acknowledgements

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